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Chemical Names And Formulas Guide

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Naming Ionic Compounds with
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Chemical Names And Formulas Guide

The common names and formulas of chemical compounds is a very relevant area of study for every competitive exam including SSC and it is important for the aspirants to develop a fair idea on this subject. In this article, we are discussing the following topics.

List of Chemical Compounds and their Common Names and Formulas

Chemical formula Synonyms CAS number
CaAl₂O₄: calcium aluminate: 12042-68-1
CaB₆: calcium boride: 12007-99-7
CaBr₂: calcium bromide: 7789-41-5
Ca(BrO)₂: calcium

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hypobromite: $\text{Ca}(\text{BrO})_2$ 2: calcium
bromite: $\text{Ca}(\text{BrO})_3$ 2: calcium bromate:
10102-75-7 $\text{Ca}(\text{BrO})_4$ 2: calcium
perbromate: $\text{Ca}(\text{CN})_2$ 2: calcium cyanide:
592-01-8 CaCO_3 3: calcium carbonate
spent lime calcite limestone marble:
471-34-1 CaC_2

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Chemical formulae provide insight into
the elements that constitute the molecules

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of a compound and also the ratio in which the atoms of these elements combine to form such molecules. For example, the chemical formula of water, which is H_2O , suggests that two hydrogen atoms combine with one oxygen atom to form one molecule of water.

Chemical Formula of Common Compounds, Chemical reaction ...
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Here is a list of ten common bases with chemical structures, chemical formulas, and alternate names. Note that strong and weak means the amount the base will dissociate in water into component ions. Strong bases will completely dissociate in water into their component ions. Weak bases only partially dissociate in water.

Names of 10 Bases With Chemical Structures and Formulas

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